

acetate (pH 2.5)-acetonitrile (87:13, v/v) as the isocratic mobile phase.

Serum. Lomefloxacin (1000 ng) in methanol (10 μ L) was added to serum samples (500 μ L) as an internal standard and then 100 μ L acetonitrile was added. The subsequent procedure was as previously described (Ohkubo et al 1992).

Calibration curves. Known amounts of ofloxacin in the ranges of 250–3000 ng g⁻¹ and 50–2000 ng mL⁻¹ were added to the blank skin tissue homogenate and the blank serum, respectively. These skin tissue and serum samples were treated according to the determination procedure described above. The peak-height ratio of ofloxacin and lomefloxacin were measured and plotted against the known concentration of ofloxacin.

Drug administration and sampling. Ofloxacin (200–300 mg as a tablet) was orally administered to patients for prevention of secondary wound infection during surgery. Venous blood samples (2.5 mL) were collected from patients (who had given informed consent) at 0.5–3 h after the administration. Sera were separated by centrifugation at 1900 g for 15 min and stored at -45°C until analysis. Skin samples, including scar tissue, haemangioma, granulation tissue, burned skin and healthy skin around the tumour were obtained from a part of the excised skin.

Results and discussion

Initially, our efforts were directed towards developing an efficient chromatographic system for the direct analysis of ofloxacin in skin using column switching techniques. In our previous paper, we studied the effects of the pH of the mobile phase on the capacity factor values of ofloxacin, norfloxacin, ciprofloxacin and enoxacin using six kinds of stationary phase (Ohkubo et al 1992). From these results, we chose enoxacin as a suitable internal standard. The separation of endogenous substance and new quinolone derivatives was satisfactory when a phenyl pre-column and an octadecyl analytical column with a 0.5% sodium acetate (pH 2.5)-acetonitrile (87:13, v/v) mobile phase system were employed (Ohkubo et al 1992). In the present paper, we studied the characterization of lomefloxacin, tosfloxacin and sparofloxacin on the six kinds of stationary phase. In all these compounds, the *k* values increased with increasing pH in

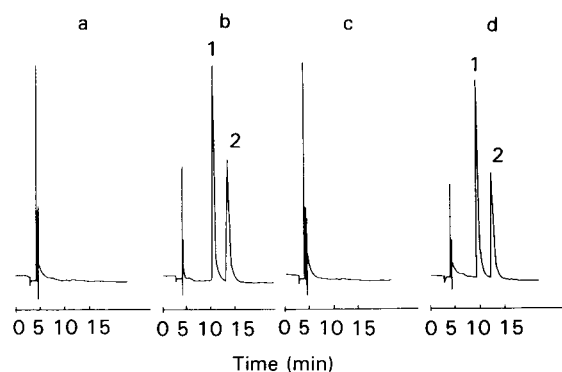


FIG. 2. Chromatograms of a: blank skin tissue homogenate, b: skin tissue homogenate in patients after treatment with 200 mg ofloxacin, c: blank serum, and d: serum in patients after treatment with 200 mg ofloxacin. 1 = ofloxacin, 2 = lomefloxacin.

Table 1. Analytical recovery and between-assay precision of ofloxacin in skin tissue homogenate.

Added (ng g ⁻¹)	Absolute recovery (%) (mean \pm s.d.)	Coefficient of variation (%)
500	84.5 \pm 3.9	4.6
1000	74.5 \pm 2.9	3.9
3000	74.3 \pm 1.6	2.2

n = 4.

the range 3.5–4.5 with 0.5% sodium acetate-acetonitrile (87:13, v/v). These quinolone derivatives were retained more strongly on phenyl, *tert*-butyl and octadecyl stationary phases. Lomefloxacin could be easily separated using these mobile phases over the entire pH range. The best resolution, peak sharpness and suitable analytical time was obtained with lomefloxacin with these stationary phases, and lomefloxacin was therefore selected as the most suitable internal standard for the determination of ofloxacin in skin tissue homogenate. A combination of a pre-column and an analytical column was selected so that interfering endogenous substances and the quinolone derivatives had different interactions with the two columns. Finally, we established a satisfactory chromatography system using lomefloxacin as an internal standard, phenyl stationary phase as a pre-column and octadecyl stationary phase as an analytical column and a 0.5% sodium acetate (pH 2.5)-acetonitrile (87:13, v/v) mobile phase system. There are several reports describing the techniques for the determination of ofloxacin in body fluids involving extraction and purification steps (Mignot et al 1988; Matsubayashi et al 1989; Okazaki et al 1991). In these reports, however, satisfactory results were not obtained due to the tedious extraction procedure and poor extraction yield. Therefore, our direct injection method with a column switching technique was satisfactory with respect to simplicity and accuracy. Extraction of ofloxacin from water and other buffer solutions was poor; in the present study, we used 0.5% sodium acetate (pH 2.5)-acetonitrile (87:13, v/v) as an extraction medium, from which we obtained a high recovery of ofloxacin from skin tissue homogenates. Fig. 2 shows representative chromatograms for a blank human skin tissue homogenate sample, a human skin tissue homogenate sample obtained from a patient treated with ofloxacin, a blank serum sample and a serum sample obtained

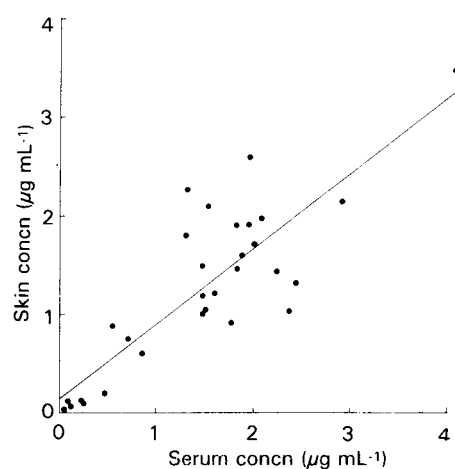


FIG. 3. Relationship between concentration of ofloxacin in skin tissue and ofloxacin in serum after oral administration of ofloxacin.

from a patient treated with ofloxacin. No significant interfering peaks derived from biological substances were seen on the chromatograms.

Calibration graphs for ofloxacin in human skin tissue and serum were linear over the ranges 250–3000 ng g⁻¹ and 50–2000 ng mL⁻¹, respectively. The limits of detection for ofloxacin were 100 ng g⁻¹ in skin tissue and 20 ng mL⁻¹ in serum. The results of recovery studies are given in Table 1. The coefficient of variation for the concentrations of ofloxacin in the patients samples ranged from 0.4 to 1.6%. The concentration of ofloxacin in skin tissue samples and serum samples was determined in patients receiving 200–300 mg ofloxacin by mouth before surgery. Fig. 3 shows that the correlation between results with skin tissue ofloxacin (y) and serum ofloxacin (x) was very good, giving the regression: $y = 0.785x + 126.9 \text{ ng mL}^{-1}$ ($r = 0.84$, $n = 30$). The average concentration ratio of ofloxacin in skin tissue vs serum was 0.84. Several workers have reported serum and skin tissue levels of ofloxacin in man (Tomizawa et al 1984; Takahashi et al 1984). However, no coefficient of correlation was obtained in this paper, because the sample number was not sufficient for statistical treatment. Our results suggest that the therapeutic basis of treatment of secondary wound-infection by oral administration of ofloxacin is firmly established in man. In conclusion, a simple, practical and accurate HPLC method has been developed for the determination of ofloxacin in human skin tissue and serum without prior extraction. The good distribution of ofloxacin from blood to skin tissue after oral administration using our method is useful for pharmacokinetic studies in skin tissue and serum in patients.

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Moment analysis of hepatic local disposition of allopurinol and oxipurinol: metabolism kinetics from allopurinol to oxipurinol in the rat isolated perfused liver

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Abstract—Drug metabolism in the liver was examined by the rat isolated perfused liver using the single-pass bolus-input technique. The test compounds, allopurinol and its metabolite oxipurinol, were independently introduced into the liver from the portal vein, and the concentration profiles in the venous outflow were monitored and kinetically analysed by moment theory. The recovery ratios of allopurinol and oxipurinol after the individual administration of each drug were estimated to be 0.17 (± 0.08 s.d.) and 1.03 (± 0.02 s.d.), respectively. The outflow recovery ratio of oxipurinol as the metabolite after allopurinol administration was estimated to be 0.80 (± 0.07 s.d.). These results indicate that the combined outflow recovery of the precursor and the metabolite after allopurinol administration is almost 100% in the rat liver.

Allopurinol and oxipurinol have been used extensively as potent inhibitors of xanthine oxidase, an enzyme that converts hypox-

anthine to xanthine, and xanthine to uric acid, for the medical treatment of hyperuricaemia and gout (Elion et al 1963; Rundles et al 1963; Elion 1966; Hille & Massey 1981). Allopurinol is metabolized by xanthine oxidase in the liver to oxipurinol (Elion et al 1966; Krenitsky et al 1967). There are very few reports on the hepatic local disposition of allopurinol and oxipurinol, the target organ of which is the liver. The purpose of this study is to evaluate in detail the local disposition of allopurinol and oxipurinol in the liver, and to elucidate the metabolic kinetics from allopurinol to oxipurinol by means of single-pass rat-liver perfusion experiments following impulse administration into the portal vein. Several models for the analysis of both local disposition and metabolism kinetics during an organ pass have been proposed of which the well-stirred model and the parallel-tube model (Pang & Rowland 1977) are the simplest. A dispersion model has also been used in the analysis of the outflow dilution curve following an impulse input (Roberts &

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